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Addition/Correction

Octamethyl-octaundecylcyclo[8]pyrrole: A Promising Sulfate Anion Extractant [*J. Am. Chem. Soc.* 2007, *129*, 11020–11021].

Leah R. Eller, Marcin Ste#pie, Christopher J. Fowler, Jeong Tae Lee, Jonathan L. Sessler, and Bruce A. Moyer

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ADDITIONS AND CORRECTIONS

Mechanistic Study of Competitive sp^3-sp^3 and sp^2-sp^3 Carbon−Carbon Reductive Elimination from a Platinum (IV) Center and the Isolation of a C−C Agostic Complex [*J. Am. Chem. Soc.* **2007**, *129*, 9538−9539]. Brian L. Madison, Summer B. Thyme, Sarra Keene, and B. Scott Williams*

Page 9538. In the seventh paragraph, second sentence, the ratio of **3**:**4** is listed as 88:12, but it should read 12:88.

Page 9539. In Scheme 2, compound **4** is depicted as having a platinum-bound triflate group, but it should instead have a methyl group, as shown in Scheme 1.

We thank Drs. Doug Wick, Luc Boisvert, and Ken Caulton for noting these errors.

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Octamethyl-octaundecylcyclo[8]pyrrole: A Promising Sulfate Anion Extractant [*J. Am. Chem. Soc.* **2007**, *129*, 11020−11021]. Leah R. Eller, Marcin Stępień, Christopher J. Fowler, Jeong Tae Lee, Jonathan L. Sessler,* and Bruce A. Moyer*

Page 11021. The reported log K'_{exch} value should be −1.1, not 4.9 as stated in the paper. The D_{SO4} values given in the text and Figure 2 remain correct. Thus, this correction does not change the basic conclusion of the paper, namely that an organic-solubilized cyclo[8]pyrrole may be used to extract sulfate effectively from a nitrate-rich aqueous solution. The value of log K'_{exch} value is still considered high as compared with that of the control, which is the simple anion exchanger acting independently. It may be noted that in the absence of **1** in Figure 1 (control system), the average value of D_{SO4} was found to be 0.0021, corresponding to log $K'_{\text{exch}}(\text{control}) = -6.0$. Referring to Figure 2, this relatively small value implies a hypothetical control D_{SO4} value of 9.7×10^{-9} for the corresponding system without the presence of **1** at 1 mM [$NaNO_3$]. By comparison, under these conditions, the measured value of D_{SO4} was 111 with **1** at 0.5 mM in the nitrate form in the organic phase. It therefore remains clear that **1** effects a remarkable enhancement relative to the feeble sulfate extraction possible by the simple anion exchanger acting independently.

Supporting Information Available: Estimation of the values of log K'_{exch} and log $K'_{\text{exch}}(\text{control})$. This material is available free of charge via the Internet at http://pubs.acs.org.

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